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House no. 5A/65, Opp. Mahual Kothi, Alpana Market, Patna Page No.: 3 14. Calculate the time in which
activity of an element reduces to 90% of its original value.

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PHYSICAL CHEMISTRY By: Shailendra Kumar Page No.: 5 5. ORDER OF REACTION AND
MOLECULARITY : Order of reaction is the sum of the power on concentration terms appearing in the rate
equation of a reaction. Molecularity of a reaction is the number of molecules involved in the "effective
collision" of the reaction.

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House no. 5A/65, Opp. Mahual Kothi, Alpana Market, Patna Page No.: 3 21) What weight of non volatile
solute, urea (NH_2CONH_2) needs to be dissolved in 100 g of water, in order to decrease the vapour
pressure of water by 25%.

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