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Aniline is an organic compound with the formula $C_6H_5NH_2$. Consisting of a phenyl group attached to an amino group, aniline is the prototypical aromatic amine. Its main use is in the manufacture of precursors to polyurethane and other industrial chemicals. Like most volatile amines, it has the odor of rotten fish. It ignites readily, burning with a smoky flame characteristic of aromatic ...

Aniline - Wikipedia

Acetic acid / $\acute{E}^{\text{TM}} \acute{E}^{\text{s}} \acute{i}\acute{E}^{\bullet} \acute{t} \acute{E}^{\text{a}} \acute{k} /$, systematically named ethanoic acid / $\acute{E}\acute{O} \acute{E}^{\text{>}} \acute{I}, \acute{E}^{\text{TM}} \acute{E}^{\text{^}} \acute{n} \acute{o}\acute{E}\acute{S} \acute{E}^{\text{a}} \acute{k} /$, is a colourless liquid organic compound with the chemical formula CH_3COOH (also written as CH_3CO_2H or $C_2H_4O_2$). When undiluted, it is sometimes called glacial acetic acid. Vinegar is no less than 4% acetic acid by volume, making acetic acid the main component of vinegar apart from water.

Acetic acid - Wikipedia

$\acute{e}^{\text{a}} \dots \acute{e}^{\text{a}} \dots \acute{e}^{\text{2}} \bullet$. IUPAC $\acute{e}^{\text{a}} \dots \acute{e}^{\text{a}} \dots \acute{e}^{\text{2}} \bullet$: $\acute{e}^{\text{,}} \acute{i}, \text{-}\acute{i}\acute{S}\text{-}\acute{i} \bullet$, $\acute{i}\text{a}, \acute{e}^{\text{,}} \acute{e}^{\text{,}} \acute{o} \acute{e} \acute{i} \acute{o} \acute{i} \bullet$, $\acute{o} \acute{e}^{\text{a}} \dots \acute{e}^{\text{a}} \dots \acute{i} \bullet \acute{e}^{\text{3}}$, $\acute{e}^{\text{,}} \acute{e}^{\text{3}} \acute{e}^{\text{'}} \acute{e}^{\text{^}} \acute{\alpha} \acute{i} \acute{z} \acute{i} \bullet \acute{E} \acute{e}^{\text{2}} \acute{f} \acute{i} \bullet$, $\acute{i}^{\text{TM}} \acute{e}^{\text{,}} \acute{o} \acute{e} \acute{i} \acute{o} \acute{e} \acute{i} \bullet$
 $\acute{i} \bullet \acute{e}^{\text{,}} \acute{o} \acute{i} \bullet \acute{i} \text{---} \acute{i} \bullet$, $\acute{o} \acute{e} \acute{i} \frac{1}{2} \bullet \acute{i} \bullet \acute{o} \acute{e}$ (alkoxy) $\acute{e}^{\text{,}} \acute{e} \acute{f} \acute{i} \acute{o} \acute{e} \frac{1}{4} \acute{e} \acute{i} \acute{o} \acute{e} \acute{e}^{\text{a}} \dots \acute{e}^{\text{a}} \dots \acute{i} \bullet \acute{o} \acute{e} \acute{e}^{\text{^}} \acute{\alpha}$. $\acute{e}^{\text{TM}} \acute{e}^{\text{,}} \frac{1}{4} \acute{i} \bullet$, $\acute{o} \acute{e} \acute{i} \frac{1}{2} \bullet \acute{i} \text{---} \bullet \text{---} \acute{i} \bullet$, $\acute{e}^{\text{,}} \acute{e} \acute{f} \acute{i} \acute{o} \acute{e} \acute{i} \bullet \acute{e}^{\text{,}} \acute{e}^{\text{TM}} \acute{i} \bullet \acute{e}^{\text{a}} \dots \acute{e}^{\text{a}} \dots \acute{i} \bullet \acute{o} \acute{e} \acute{e}^{\text{^}} \acute{\alpha}$; $\acute{e}^{\text{^}} \acute{E} \acute{i} \acute{S} \acute{o} \acute{e}^{\text{a}} \dots$: $\acute{e}^{\text{'}} \acute{e}^{\text{^}} \acute{o} \acute{e} \acute{i} \bullet$
 $\acute{i} \bullet \acute{O} \acute{E} \acute{i}, \text{-}\acute{e}^{\text{,}} \acute{o}$ (alkyl group) $\acute{e}^{\text{^}} \frac{1}{4} \bullet \acute{O} \acute{E} \acute{i} \acute{O} \acute{E} \acute{e}^{\text{23}} \acute{i} \acute{o} \acute{e} \acute{i} \bullet$, $\acute{o} \acute{e} \acute{O} \acute{E} \acute{e} \acute{i} \acute{o} \acute{e} \acute{e}^{\text{a}} \dots \acute{e}^{\text{a}} \dots \acute{i} \bullet \acute{e}^{\text{3}}$, $\acute{e}^{\text{'}} \acute{i} \text{---} \bullet \text{---} \acute{i} \bullet$, \acute{o} (ether) $\acute{e}^{\text{^}} \frac{1}{4} \dots$

$\acute{i} \text{---} \acute{i} \bullet$, \acute{o} - $\acute{i} \acute{o} \acute{e} \dots \acute{i} \bullet \acute{e}^{\text{^}} \pm \acute{e}^{\text{3}} \frac{1}{4}$, $\acute{i} \acute{S} \acute{o} \acute{e} \acute{i} \text{---} \acute{e}^{\text{a}} \acute{e}^{\text{'}} \acute{o} \acute{e} \acute{i} \bullet \acute{e}^{\text{^}} \pm \acute{e}^{\text{3}} \frac{1}{4} \acute{i} \bullet \text{---} \acute{i} \bullet$,,

Jako Markovnikovovo pravidlo (nĂkdy tĂn sprĂvnĂ nazĂvĂno markovnikovo) se v organickĂ chemii oznaĂuje pozorovĂnĂ zaloĂenĂ na pravidle ZajcevovĂ. Formuloval ho ruskĂ chemik Vladimir VasiljeviĂ Markovnikov v roce 1870.. Pravidlo Ă-kĂ, Ă pĂmidĂnĂ-m protickĂ kyseliny HX k alkenu se vodĂ-k kyseliny navĂjĂe na atom uhlĂku s mĂnĂ alkylovĂmi substituenty, kdeĂto ...

Markovnikovovo pravidlo â€“ Wikipedie

History: Kilgore College is a publicly supported, two-year, comprehensive community college offering postsecondary educational opportunities. In 1935 Kilgore College was the idea of Mr. W. L. Dodson and the community of Kilgore, Texas.

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