

**biochemistry voet 4th edition pdf**

Ein Kohlenhydrat besitzt die allgemeine Summenformel  $C_n H_{2n} O_n$  mit  $n \geq 3$ . Alternativ können Kohlenhydrate definiert werden als Polyhydroxyaldehyde und -ketone und Moleküle, die nach Hydrolyse solche ergeben – allerdings werden auch deren Abkömmlinge wie Desoxyribose als Kohlenhydrate bezeichnet, welche eine andere Summenformel aufweisen. Die IUPAC definiert Kohlenhydrate ...

**Kohlenhydrate – Wikipedia**

A non-covalent interaction differs from a covalent bond in that it does not involve the sharing of electrons, but rather involves more dispersed variations of electromagnetic interactions between molecules or within a molecule. The chemical energy released in the formation of non-covalent interactions is typically on the order of 1-5 kcal/mol (1000–5000 calories per  $6.02 \times 10^{23}$  molecules).

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